

Morphology of Oligomers and Their Network Polymers

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SUMMARY

The morphology of di(4-methacryloxyethylene carbamate toluylene-2-carbamic) esters of oligoethylene glycols (OUM) with different molecular weight, of their network polymers and of model oligomers without methacrylic groups was studied. The supermolecular structure of oligomers is found to depend on molecular weight. Globular structure is typical for low molecular weight oligomers with $\bar{M}_n \approx 1200$. The structure of crystalline samples with $\bar{M}_n = 1600-3600$ is lamellar. The latter is retained on melting. It is transformed into globular structure at a temperature much higher than the melting point. A correlation exists between the morphology of polymerizable oligomers and their network polymers.

The process of network formation from oligomers is extremely complicated. Irrespective of the mechanism it is influenced by MWD, FTD, intermolecular interaction and a number of other factors. The interconnection between the morphology of polymerizable oligomers and the supermolecular structure of their crosslinked polymers has not been sufficiently studied. Information about this connection facilitates phenomenological understanding of the network formation process.

EXPERIMENTAL

The OUM's and model compounds (OUE) were prepared by reaction of different molecular weight oligoethylene glycols with 2-isocyanate toluylene-4-methacryloxyethylene (or 4-ethyl) urethane (Tabl.1) (ZEMSKOVA et al. 1973); network polymers were obtained by radical polymerization of OUM's (MATVEEVA et al. 1974). The supermolecular structure of liquid oligomers and melts was fixed by fast freezing of samples with liquid nitrogen; micrographs of replicas of oligomer and polymer samples were obtained with an EUMB-100B electron microscope; the fracturing of glassy oligomers and deposition of the carbon film on the fracture surface were carried out under vacuum at $-150 - -160^\circ\text{C}$ (MATVEEVA,

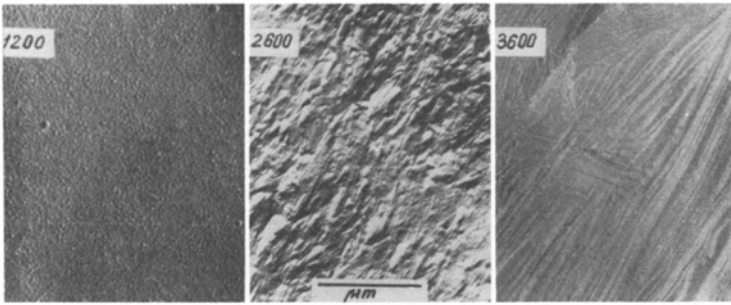


Fig.1. Morphology of different molecular weight oligomers

oligomer morphology on the temperature was studied in order to estimate the stability of a supermolecular organization of oligomers and to find the correlation

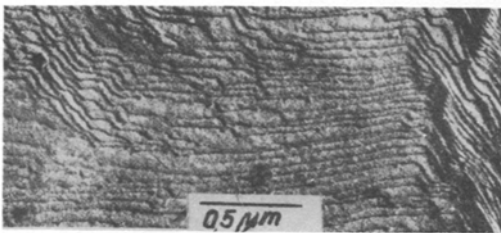


Fig.2. Morphology of oligomers with $M_n = 2250$ in a liquid-crystalline state.

between the morphology of oligomers and their network polymers. Model oligomers without methacrylic groups were synthesized to exclude the polymerization of OUM's. Micrographs of fracture replicas of vitrified oligomer melts heated to $\sim 10^\circ$ and $\sim 100^\circ$ above the melting points are demonstrated in Fig.3.

The thickness of lamellae of vitrified melts is close to the values typical for the same oligomers in the crystalline state. With further increase of the temperature, globules emerge on the surface of the lamellae. The lamellar structure is completely transformed to a globular one at a temperature much higher than the melting point (MATVEEVA, KISELEV et al. 1979).

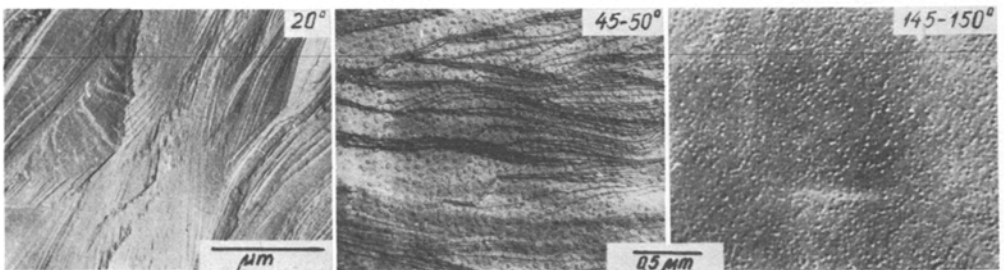


Fig.3. Temperature transformation of the morphology of the oligomer with $M_n = 3400$.

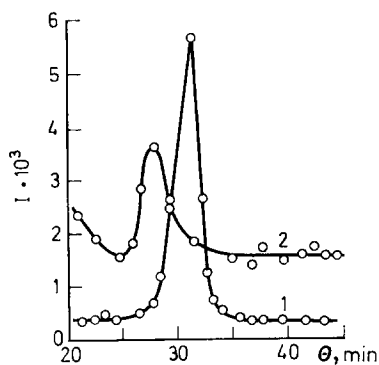


Fig.4. Small-angle X-ray scattering curves for oligomers with $\bar{M}_n = 1200$ (2) and $\bar{M}_n = 2250$ (1).

Small-angle X-ray scattering curves display a distinct maximum which indicates the presence of oligomer organization at the supermolecular level. The position of the maximum varies with the molecular weight of the oligomer (Fig.4). The value of the large period for the oligomer with $\bar{M}_n = 1200$ is 150\AA and for the oligomer with $\bar{M}_n = 2250$, 170\AA . The results of the small-angle X-ray scattering study of oligomer morphology confirm the electron microscopy data.

The supermolecular organization of the mixtures of oligomers is controlled by the molecular weight and the ratio of initial components. The presence of lamellar structure (thickness $\sim 100\text{\AA}$) decorated by globules of the low molecular oligomer ($d \approx 100\text{\AA}$) is characteristic of the mixture of the OUE's with $\bar{M}_n = 600$ and 3400 enriched by the high molecular component (weight ratio 1:20) (Fig.5a). The morphological pattern essentially varies when a part of the low molecular weight oligomer increases. Separate lamellar incorporations typical of the high molecular ingredient are observed for the mixture of components taken in equal parts by weight. At the same time the background of the sample consists of globules (homogeneous in size) of the low molecular weight oligomer (Fig.5b).

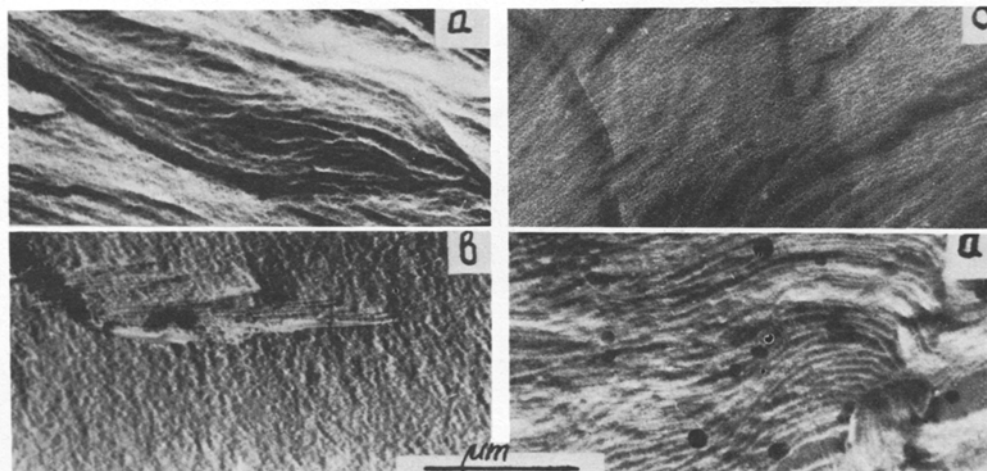


Fig.5. The morphology of mixtures of oligomers as a function of molecular weights and composition.

The increase in molecular length of the low molecular part of the mixture ($\bar{M}_n=1040$ and 3400) practically do not influence the morphology of the high molecular weight component at a small content of the former (weight ratio 1:20). In this case the distinctly formed lamellar structure is retained; the thickness of lamellae is close to the value for the crystalline state of this component; however, accumulation of the globules and erosion of the lamella edges in individual separate regions are observed (Fig.5c). The variation in the oligomer ratio in the mixture towards the low molecular weight component (weight ratio 1:1) leads to essential looseness of the structure - the thickness of the lamellae increases and the sharpness of the lamella edges disappears (Fig.5d).

Thus, the morphology of the oligomers is controlled by their molecular weight. The studied oligomers are characterized by the chain regularity, by the terminal groups capable to form H-bonds and by the narrow MWD. All that ensure preservation of the order in the oligomer melts at temperatures higher than the melting points. Fractionation in size of the macromolecules is observed for mixtures of different molecular weight oligomers prepared by mixing of the solution of the components and by subsequent evaporation of the solvent. As a result the morphological pattern represents a combination of the supermolecular elements characteristic of the initial oligomers.

There exists a connection between the supermolecular structure of crosslinked polymers and the morphology of initial oligomers. Network polymers prepared by curing low molecular weight oligomers are composed of globules and the different formations consisting of coalesced globules. The morphological pattern of polymers obtained from oligomers having a lamellar structure does not differ from the structure of the initial oligomers. The lamellar structure is observed also for an amorphous polymer with the value of interstitial chains $\bar{M}_c = 1600$ (Fig.6).

The supermolecular structure network polymer obtained by curing the mixture of oligomers with different molecular weight is characterized by the presence of various structural elements. Regions of globular and "spherulite"-like formations have been found in an amorphous polymer based on the mixture of oligomers with $\bar{M}_n=670$ and $\bar{M}_n=6600$ taken in equal parts by weight (Fig.7a,b). A polymer obtained by curing the mixture of equal parts by weight of OUM's with $\bar{M}_n = 750, 1600$ and 3600 is amorphous and has a globular-lamellar structure with eroded boundaries after two-week standing at room temperature. After more than a three-year standing this polymer crystallizes to a small degree of crystallinity (<10%) and its structure remains

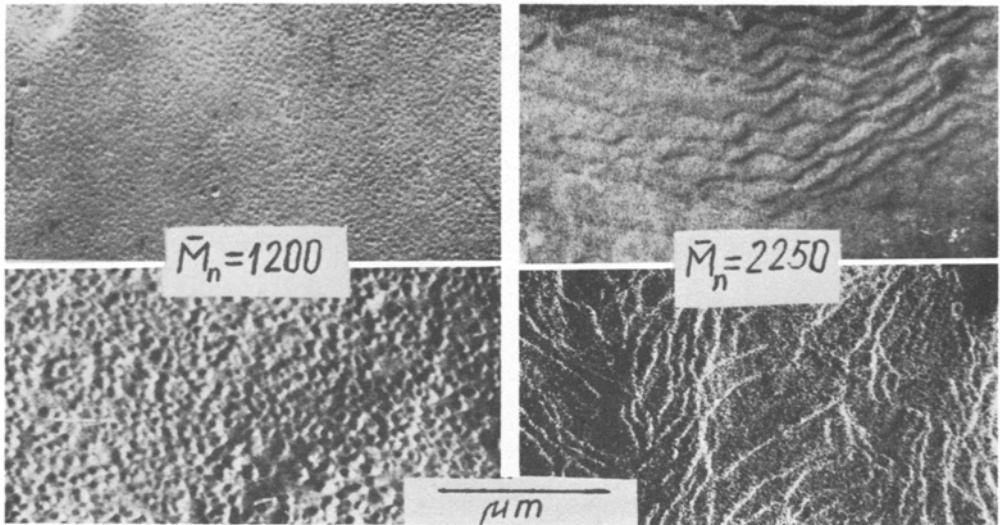


Fig.6. Correlation between morphology of oligomers and their network polymers.

globular, but the lamellae become more distinct (Fig. 7c). A similar phenomenon is observed for the polymer prepared from a mixture consisting of eight oligomers with narrow MWD and different molecular weight ranging from $M_n = 670$ to $\bar{M}_n = 6600$. After two-week standing this polymer is amorphous and has a globular structure. After long standing it partly crystallizes (<10%); rather well formed lamellae are present in its structure together with globular formations.

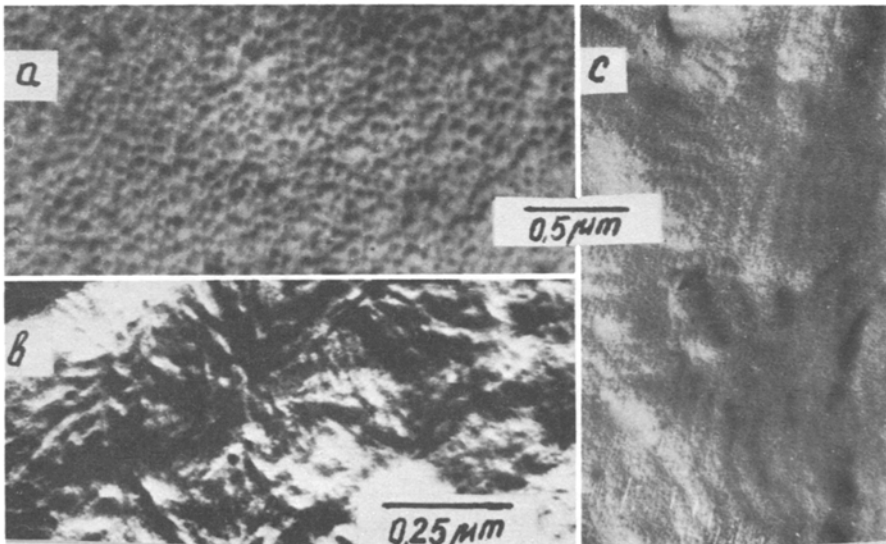


Fig.7. Morphology of network polymers based on mixtures of oligomers.

The following phenomenological pattern of the formation of crosslinked polymers by polymerization of α, ω -bis unsaturated oligomers may be suggested (BERLIN 1977; MATVEEVA 1977):

During crosslinking polymerization of low molecular weight oligomers the resulting primary polymethacrylic chains, having an oligomeric pendant in every unit, coil and aggregate with each other. The growth of macrochains continues due to consumption of the oligomer immobilized in the coils and of the methacrylic groups in the pendant oligomer blocks. The macrochains branch out, form cycles, and crosslink together. As a result, stabilization of these aggregates takes place. In the course of this process the density of aggregate particles gradually increases as a result of their contraction. Simultaneously the syneresis of the soluble polymerizate may take place. The individual particles become linked together by polymethacrylic tie-chains and oligomeric blocks. When the number of such particles and of the branched and tie-chains becomes sufficient for immobilizing the liquid reactant, there occurs gelation, and the system loses its fluidity. As a result, a microheterogeneous three-dimensional crosslinked system is formed and highly crosslinked microparticles are interconnected by zones having an essentially lower crosslinking density.

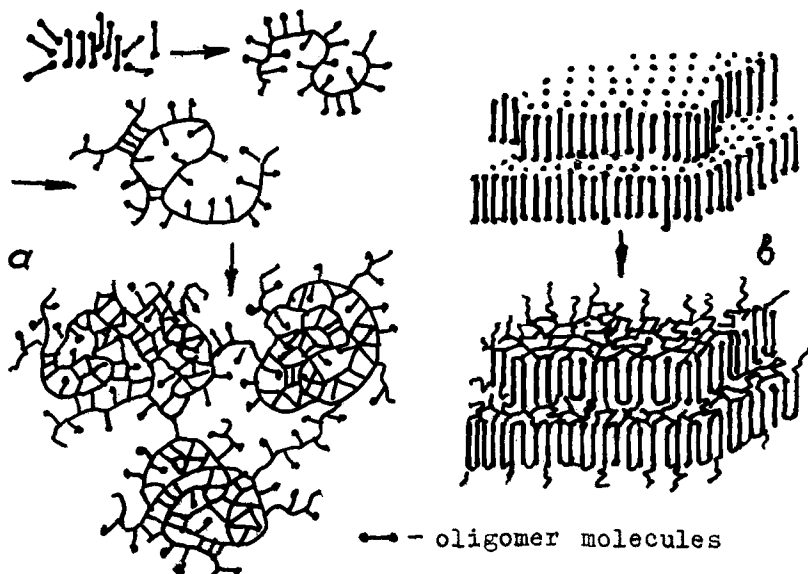


Fig.8. Scheme of the network formation from oligomers.

In the case of crystalline oligomers their transition on melting to the liquid-crystalline state with lamellar structure is observed. In such systems the polymerization process develops in the interlayer zones where terminal functional groups are concentrated. The mobility of the terminal groups of oligomeric molecules in these zones is quite sufficient for realization of the process of propagation and conformational transformations of growing primary macrochains. As a result of three-dimensional polymerization, the terminal groups of oligomeric molecules become cross-linked and lamellar structure of the liquid is fixed in the polymer.

The process of the network formation in the case of oligomeric mixtures includes elements of the mechanisms described above. The supermolecular structure of such polymers is either completely globular, or it consists of combinations of globular formations with lamellae not well defined or of other structural elements. However, in spite of considerable disorder and the limited mobility of interstitial chains, a gradual increase of order in the system, a crystallization of long-chain oligomeric blocks, and a formation of lamellar structure regions may be observed in the network polymer.

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